

Effect of Hydrostatic Pressure on the Phase Transitions in Ferroelectric RbHSO₄ and RbDSO₄

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Effect of hydrostatic pressure on the ferroelectric Curie temperatures of RbHSO₄ and RbDSO₄ was studied by dielectric constant measurements at high pressures. The observed pressure coefficients of the Curie temperature were 12.0 ± 0.1 deg kbar⁻¹ for RbHSO₄ and 12.7 ± 0.2 deg kbar⁻¹ for RbDSO₄. A pressure-induced phase was found for RbHSO₄ above about 7.5 kbar.

§1. Introduction

Rubidium bisulfate RbHSO₄ is ferroelectric below a Curie temperature of -9.4°C .^{1)*} The crystal structures of the paraelectric phase (denoted as Phase I in this paper) and the ferroelectric phase (Phase II) are respectively considered to be isomorphous with the higher temperature paraelectric phase (Phase I) and the ferroelectric phase (Phase II) of NH₄HSO₄.^{1,2)} Therefore, a common mechanism would be invoked for the I-II transitions in these two bisulfates. A marked difference in the phase transitions of these two bisulfates is that NH₄HSO₄ is ferroelectric between two Curie temperatures while RbHSO₄ does not show a lower Curie temperature down to liquid helium temperature.³⁾ These compounds show a peculiar isotope effect for substituting hydrogens by deuteriums; the deuterization depresses the Curie temperature (the upper Curie temperature for the case of NH₄HSO₄) unlike other hydrogen containing ferroelectrics.^{4,5)}

Bridgman⁶⁾ studied the pressure-temperature phase diagram of NH₄HSO₄ above room temperature and found that the room temperature phase (Phase I) transforms to a high pressure phase (Phase II_B) at about 0.5 kbar at room temperature. The phase boundary between Phases I and II_B extrapolates itself to the lower Curie temperature at 0 kbar. Although it is somewhat questionable that Phase II_B is identical with the low temperature non-ferroelectric

phase of Phase III,⁷⁾ the ferroelectric phase of NH₄HSO₄ is restricted in a rather narrow region in the p - T plane.⁸⁾

The present work intended to draw the pressure-temperature phase diagram of RbHSO₄ in order to decide whether the ferroelectric state of RbHSO₄ is also restricted in a limited region in the p - T plane. A part of the present work was published.⁹⁾ The present paper reports in a wider pressure region up to about 8 kbar. A first order phase transition has been observed above about 7.5 kbar. The pressure dependence of the Curie temperature of RbDSO₄ was also determined to compare the results of RbHSO₄.

§2. Experimental

Single crystals of RbHSO₄ were prepared by means of slow evaporation of an aqueous solution at room temperature. A single crystal of RbDSO₄ was kindly provided by Dr. I. Tatsuzaki. Several c -plates were cleaved out of a single crystal. After attached silver-paste electrodes, the specimen was set in a high pressure bomb. Two types of pressure bomb were used; one was an internal furnace type for high temperature measurements, and the other was a Cu-Be one for low temperatures. Each bomb has seven electrical terminal plugs. Kerosene was used as the pressure transmitting fluid. The dielectric constant was measured at 100 kHz with an LCR meter and recorded on an x-y recorder. Detailed descriptions of the high pressure system and the procedure of dielectric measurements at high pressures were reported elsewhere.^{10,11)}

§3. Results

Figure 1 shows the temperature dependence

* Pepinsky *et al.*¹⁾ reported the Curie temperature of RbHSO₄ to be about -15°C , but more recent studies showed that it is around -10°C . The value -9.4°C was obtained by Chihara³⁾ from a heat capacity measurement.

of the dielectric constant along the polar direction of the *c*-axis at various hydrostatic pressures. The dielectric constant peak shifts towards high temperatures as pressure increases. At 6.85 kbar, immediately after reaching the

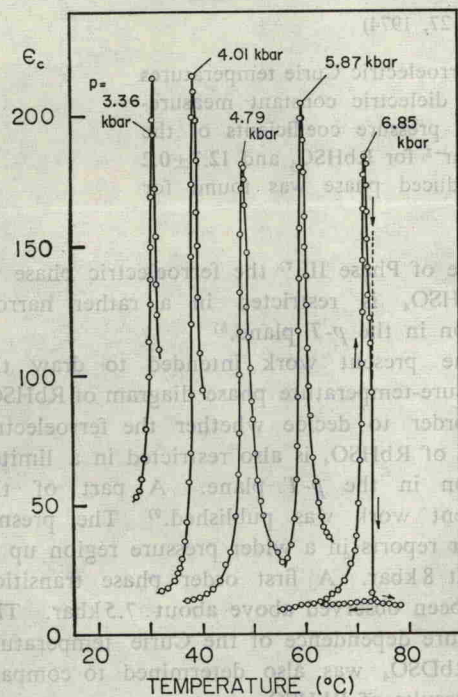


Fig. 1. Temperature dependence of the dielectric constant ϵ_c along the *c*-axis of RbHSO_4 at various hydrostatic pressures. For the curve at $p=6.85$ kbar, the transition from Phase II to Phase III is indicated by the vertical slashed line and arrow. Frequency: 100 kHz.

maximum at the Curie temperature, the dielectric constant discontinuously decreases and it is not recovered even if temperature was lowered far below the corresponding Curie temperature. The discontinuous change in dielectric constant evidently corresponds to a first order phase transition to a pressure-induced phase which is denoted as Phase III in this paper. Figure 2 indicates a isothermal variation of the dielectric constant with pressure at 36°C. At about 7.5 kbar, a discontinuous change accompanying the II-III transition is seen. The decrease in dielectric constant at the II-III transition was not recovered by lowering pressure to 1 atm. The fact indicates that the high pressure III Phase can be brought to atmospheric pressure as a metastable structure.

The resultant pressure-temperature phase diagram of RbHSO_4 is illustrated in Fig. 3. The ferroelectric Curie temperature linearly increases with increasing pressure with a slope of $dT_c/dp=12.0\pm 0.1$ deg kbar⁻¹. The Curie temperature at 1 atm was $-(9.5\pm 0.1)^\circ\text{C}$ which is in good agreement with $-(9.4\pm 0.1)^\circ\text{C}$ obtained by a recent heat capacity measure-

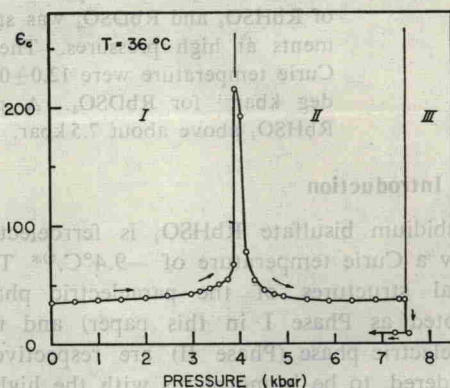


Fig. 2. Pressure dependence of the dielectric constant ϵ_c along the *c*-axis of RbHSO_4 at $T=36^\circ\text{C}$. Frequency: 100 kHz.

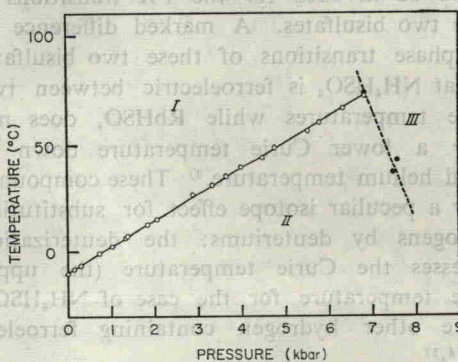


Fig. 3. Pressure-temperature phase diagram of RbHSO_4 . (The phase diagram of NH_4HSO_4 above mentioned by Chihara.³) Since the specimen was cracked when it was once brought into Phase III, the equilibrium II-III (or I-III) phase boundary was not determined. In Fig 3, an estimated II-III phase boundary is represented only for increasing pressure run (or increasing temperature run) by a slashed line. The I-II-III triple point is roughly estimated to be at $T=73^\circ\text{C}$, $p=6.9$ kbar.

The pressure dependence of the Curie temperature of RbDSO_4 is shown in Fig. 4. The pressure coefficient of the Curie temperature of RbDSO_4 was determined as $dT_c/dp=12.7$